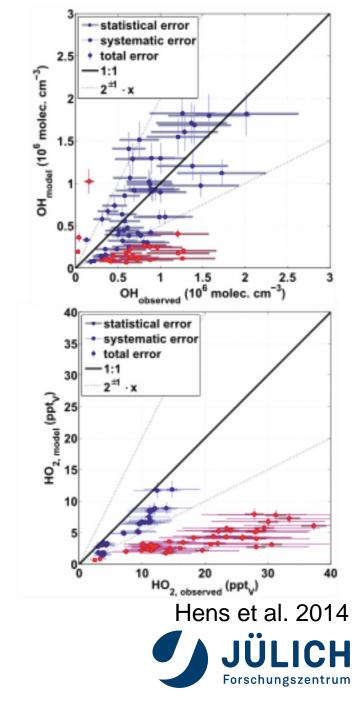
# Investigation of the $\alpha$ -pinene & $\beta$ -pinene photooxidation by OH in the atmospheric simulation chamber SAPHIR

Michael Rolletter, Martin Kaminski, Ismail-Hakki Acir, Birger Bohn, Theo Bauers, Hans-Peter Dorn, Rolf Häseler, Frank Holland, Xin Li, Anna Lutz, Sacha Nehr, Franz Rohrer, Ralf Tillmann, Luc Vereecken, Robert Wegener, Andreas Hofzumahaus, Astrid Kiendler-Scharr, Andreas Wahner and Hendrik Fuchs

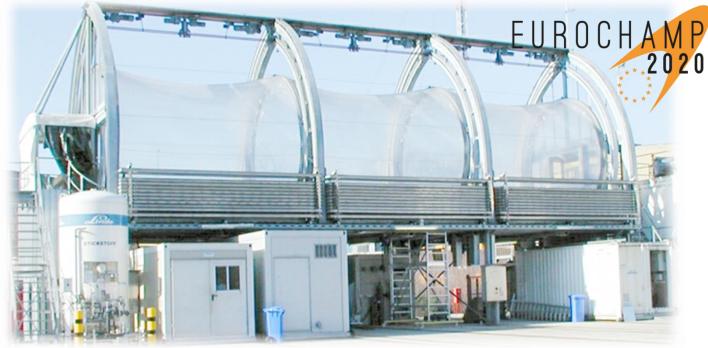


#### **Motivation**

- High emission rates and OH reactivity of monoterpenes in forested areas
- Large discrepancies between measured and modeled OH concentrations in monoterpene dominated environments (e.g. Kim et al., ACP 2013; Hens et al., ACP 2014)



### **Atmospheric simulation chamber SAPHIR**



- Volume 270 m<sup>3</sup>
- Walls made of Teflon film (high UV transmission)
- Light source: solar radiation
- Photolysis frequencies calculated based on actinic flux measurements

#### **Radical source**

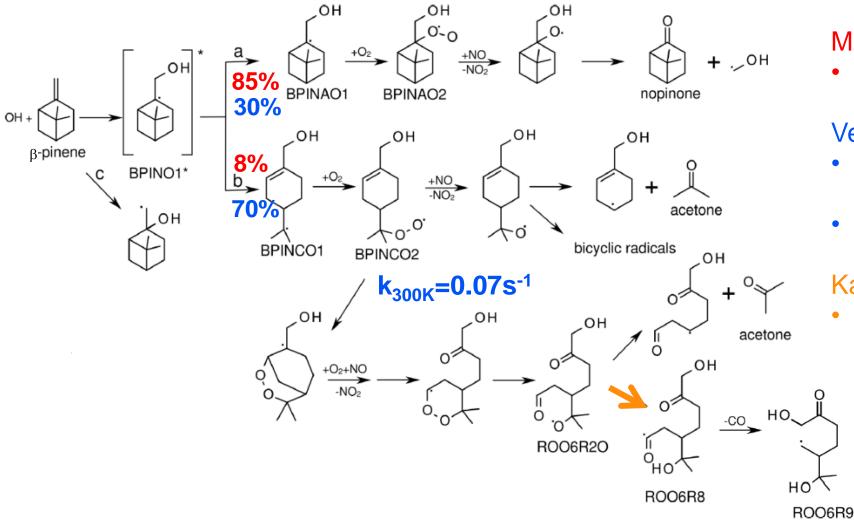
Wall	+ $hv$	$\rightarrow$ HONO(g)	
HONO(g) + hv		$\rightarrow$ OH + NO	

#### Instrumentation:

- OH, HO<sub>2</sub>, k(OH): Laser-induced fluorescence (LIF)
- OH: Differential Optical Absorption (DOAS, absolute technique)
- $\alpha$  &  $\beta$ -pinene, pinonaldehyde, nopinone, acetone: GC-FID and PTR-TOF-MS,
- HCHO: (Hantzsch, DOAS)
- HONO: Long Path Absorption Photometer (LOPAP)
- NO, NO<sub>2</sub>: Chemiluminescence detectors
- O<sub>3</sub>: UV Absorption



#### **β-pinene degradation – MCM and modifications**



MCM 3.2

HO,

нo

Channel a) favoured

Vereecken & Peeters, PCCP, 2012

- Channel b) with ring-opening favoured
- Additional ring-closing step

Kaminski et al. ACP, 2017

Additional HO<sub>2</sub> formation

-HO<sub>2</sub>

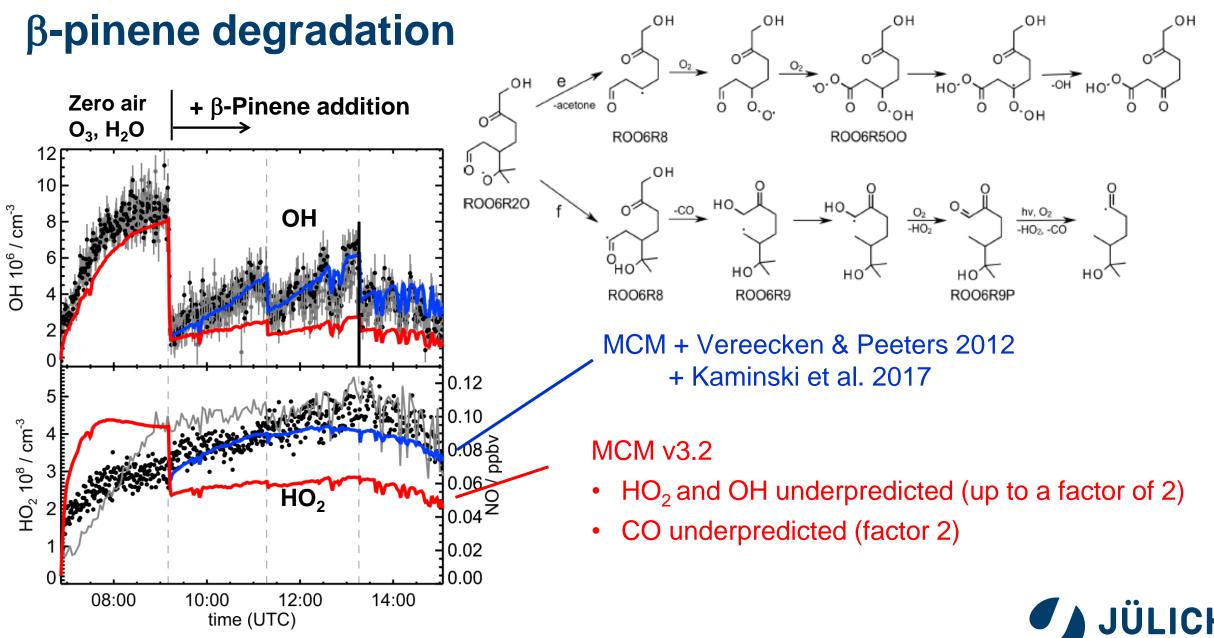
HO

ROO6R9P



hv, O<sub>2</sub> -HO<sub>2</sub>, -CO

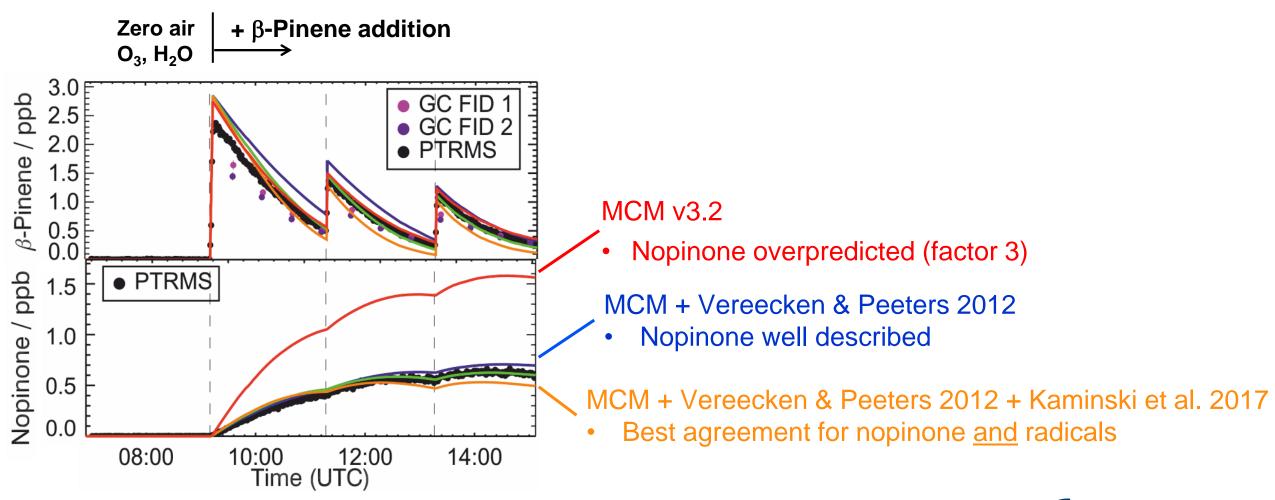
HO



Forschungszentrum

Mitglied der Helmholtz-Gemeinschaft

## **β-pinene degradation**





### **β-pinene results**

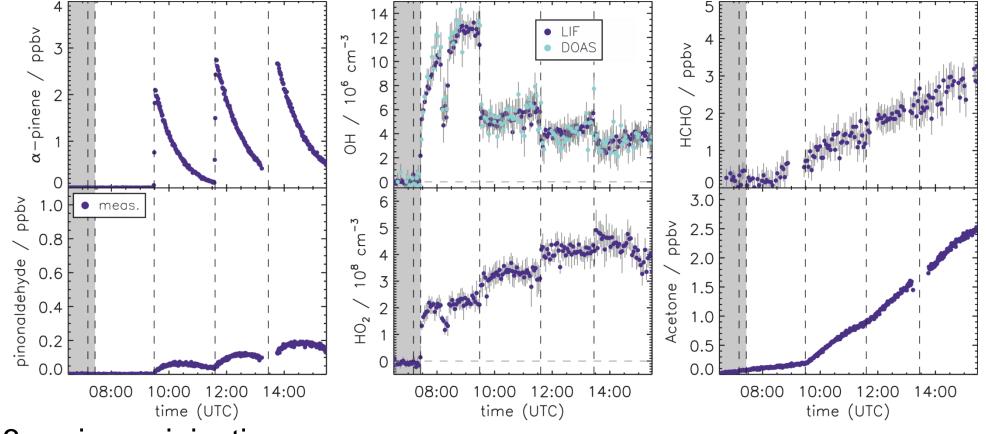
- OH and HO<sub>2</sub> concentrations are underestimated by MCM
- Additional HO<sub>2</sub> increases OH by radical recycling ("closed OH budget")
  - $\rightarrow$  Similar findings in field campaigns
- Measured nopinone yields is smaller than the MCM predicts

 $\rightarrow$  Additional RO<sub>2</sub> chemistry suggested by Vereecken et al. improves model-measurement agreement (radicals and organic products)

 $\rightarrow$  Adjustment of branching ratios and assumption additional HO<sub>2</sub> production would explain observations



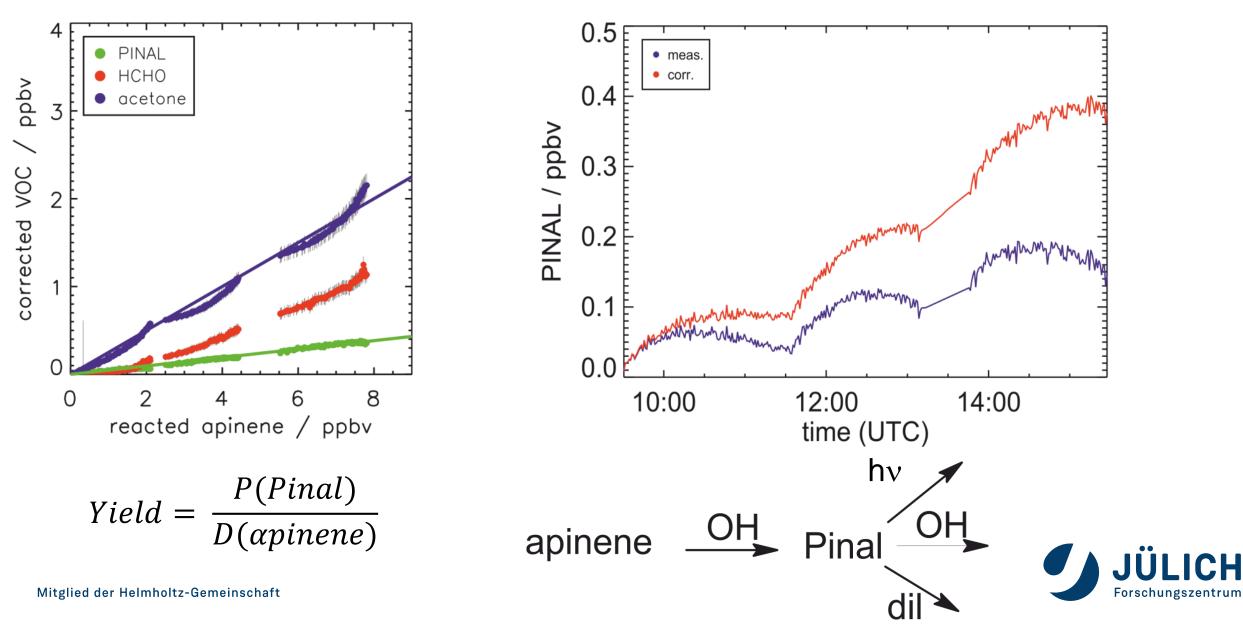
#### $\alpha$ -pinene degradation



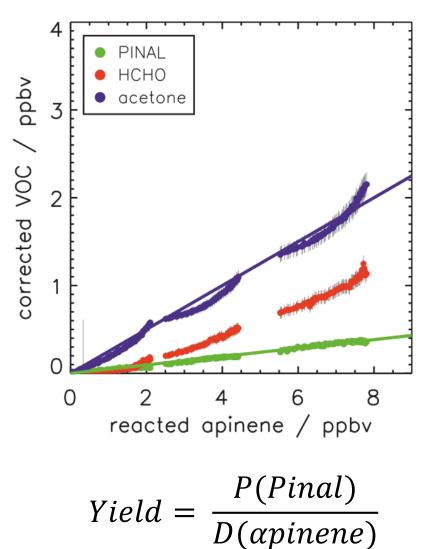
- $3 \alpha$ -pinene injections
- NO < 100 ppt
- Atmospheric O<sub>3</sub> level
- RH= 30-60 %



#### **Product yields**



#### **Product yields**



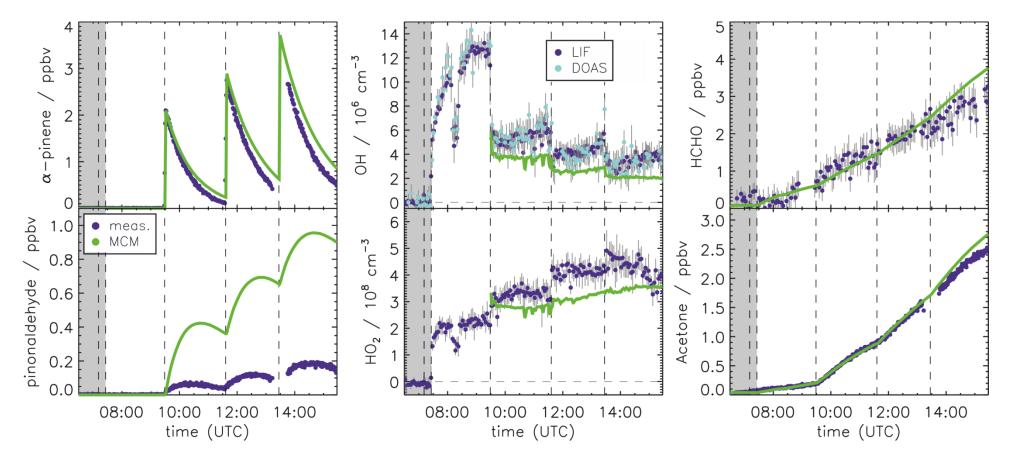
- PINAL yield:  $8 \pm 3 \%$
- HCHO yield:  $5 \pm 1 \%$
- Acetone yield:  $24 \pm 2\%$

	Yield / %			Exp. cond	Exp. conditions	
Reference	Pinonaldehyde	Acetone	CH2O	α-pinene / ppb	NO/ppb	Water / rH %
Arey et al. (1990)	29	-	-	400-900	10000	0
Hakola et al. (1994)	$28 \pm 5$	-	-	350-1000	10000	0
Noziere et al. (1999)	$87\pm20$	$9\pm 6$	$23\pm9$	200-2700	4000	0
Larsen et al. (2001)	$6\pm 2$	$11\pm 3$	$8\pm1$	1400-1600	1000	2-5
Aschmann et al. (2002)	$28 \pm 5$	-	-	400-900	7000-9000	0
Lee et al. (2006)	$30 \pm 0.3$	6	16	109	9	0
Eddingsaas et al. (2012)	27-35	<10	-	20-50	high NO	<10
Noziere et al. (1999)	$37 \pm 7$	$7\pm2$	$8 \pm 1$	200-2700	NO free	0
Wisthaler et al. (2001)	$34 \pm 9$	$11 \pm 2$	$8 \pm 1$	1000-1300	NO free	0
Eddingsaas et al. (2012)	20	-	-	20-50	low NO	<10
this work	$8\pm3$	$24\pm2$	$5\pm1$	3.8	< 0.1	30-60

 $\rightarrow$  Lack of studies under realistic atmospheric conditions



#### $\alpha$ -pinene degradation – MCM 3.3.1



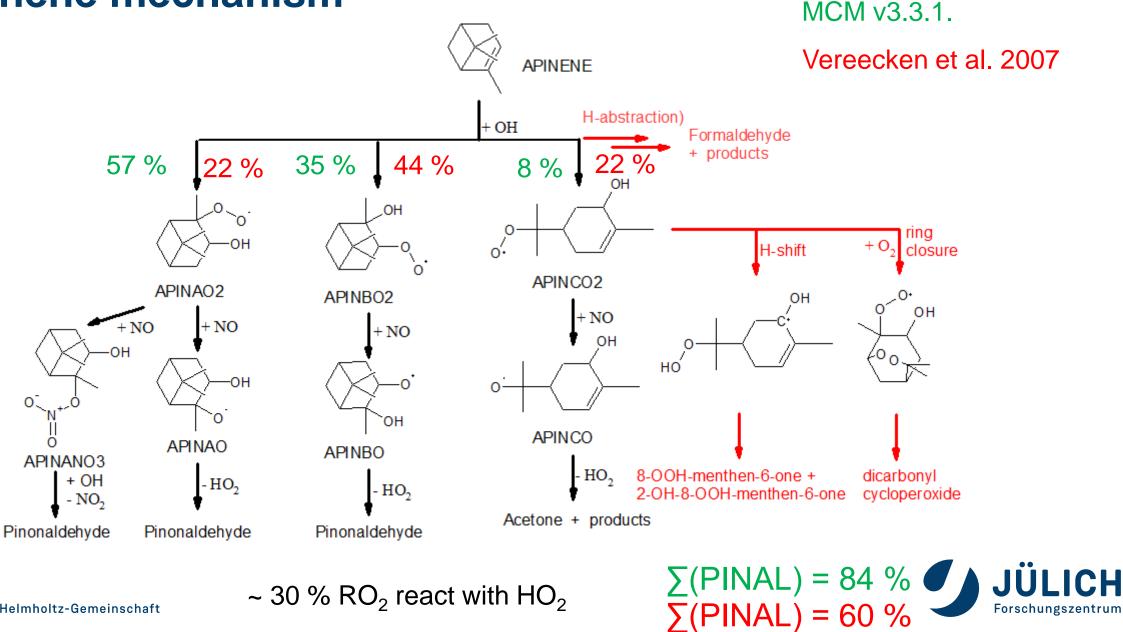
MCM 3.3.1 results:

- Pinonaldehyde production is overestimated by a factor of 4
- OH and  $HO_2$  are underestimated by 25 %

Mitglied der Helmholtz-Gemeinschaft



#### $\alpha$ -pinene mechanism

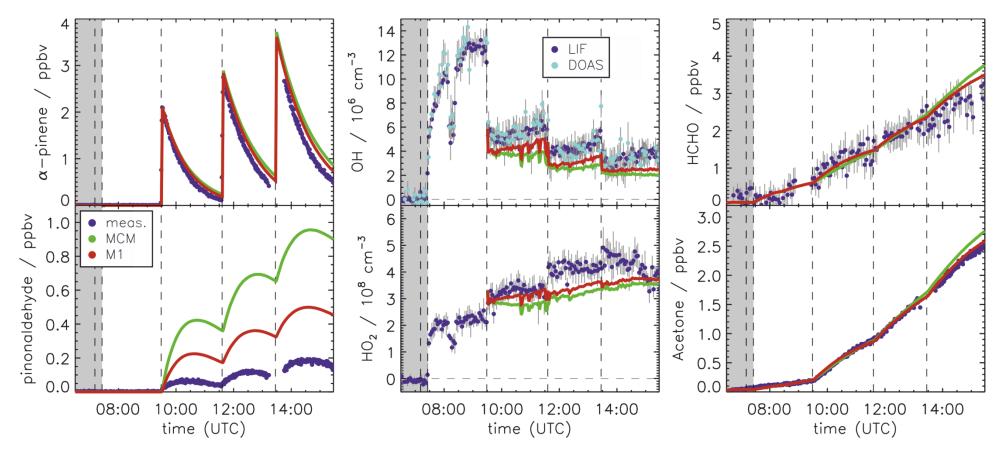


Forschungszentrum

Mitglied der Helmholtz-Gemeinschaft

 $\sim 30 \% \text{ RO}_2$  react with HO<sub>2</sub>

#### $\alpha$ -pinene degradation – Vereecken et al.

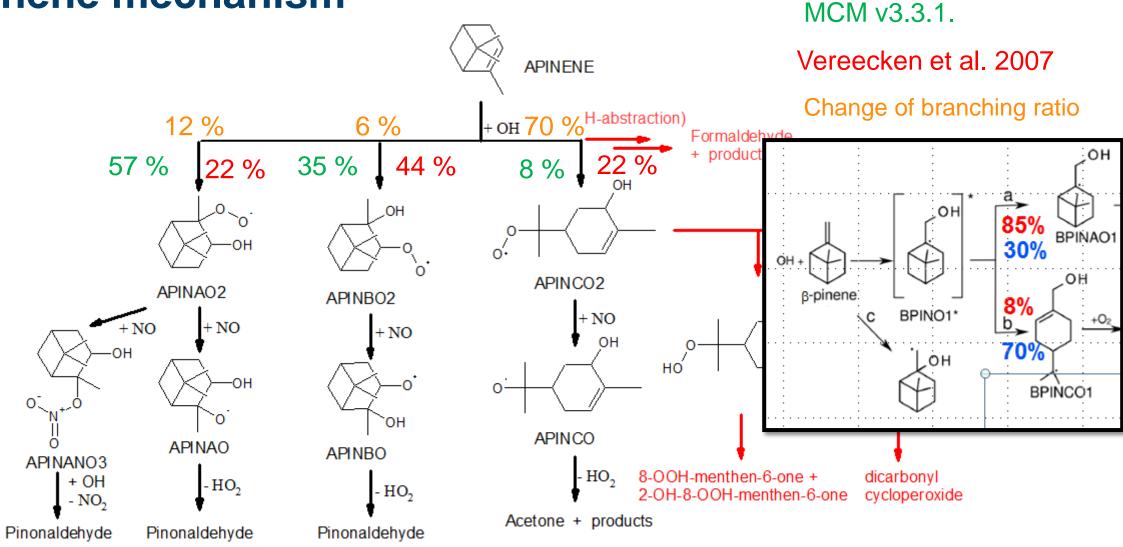


Vereecken mechanism results:

- Pinonaldehyde production is still overestimated by a factor of 2
- OH and  $HO_2$  are still underestimated by 25 %

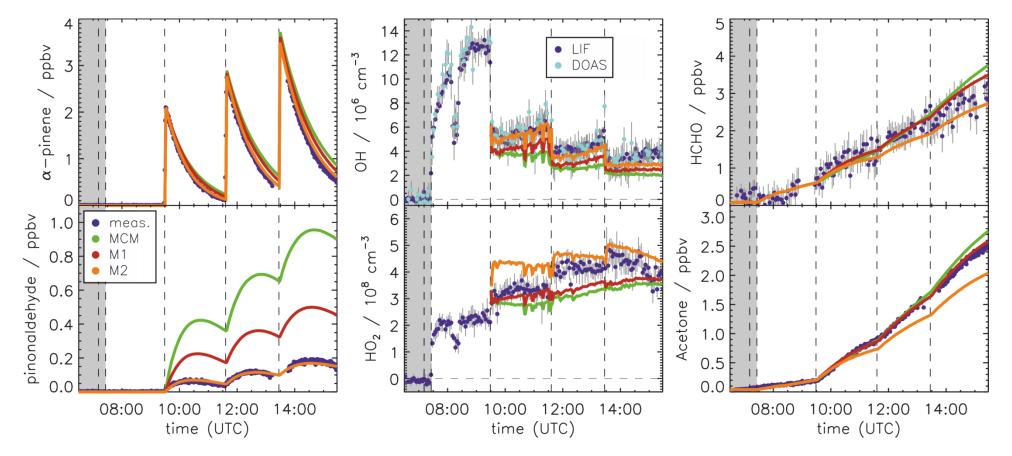


#### $\alpha$ -pinene mechanism





#### $\alpha$ -pinene degradation – modified Vereecken et al.



Change in branching ratio improves OH agreement between measurement and model



#### Conclusions

- OH and HO<sub>2</sub> concentrations are underestimated by models
- Measured yields of major aldehyde products (pinonaldehyde and nopinone) are smaller than the MCM predicts
- →Additional RO<sub>2</sub> chemistry suggested by Vereecken et al. improves model-measurement agreement (radicals and organic products)
- → Adjustment of branching ratios for RO<sub>2</sub> isomers from OH +  $\alpha$ -pinene can bring model prediction into agreement for  $\alpha$ -pinene

Further studies (theo. & exp.) are necessary to resolve the detailed  $\alpha$ -pinene chemistry under atmospheric conditions



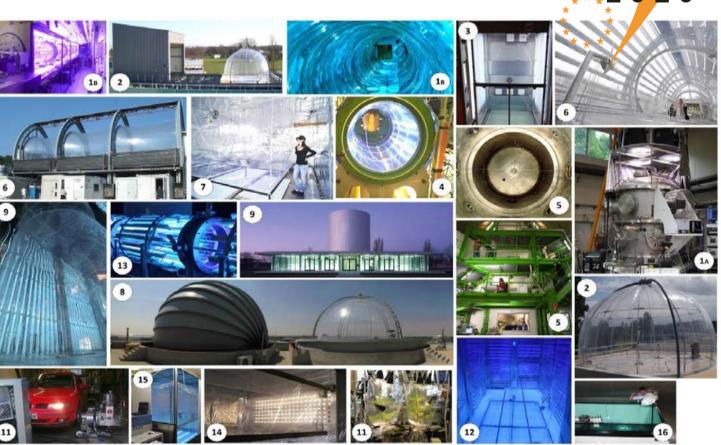
## TRANSNATIONAL ACCESS TO EUROPEAN ENVIRONMENTAL CHAMBERS



Eurochamp offers:

- Free access to chambers
- Travel support
- → Application: <u>www.eurochamp.org</u>











Mitglied der Helmholtz-Gemeinschaft

## **β-pinene degradation**

Kaminski et al., ACP 2017

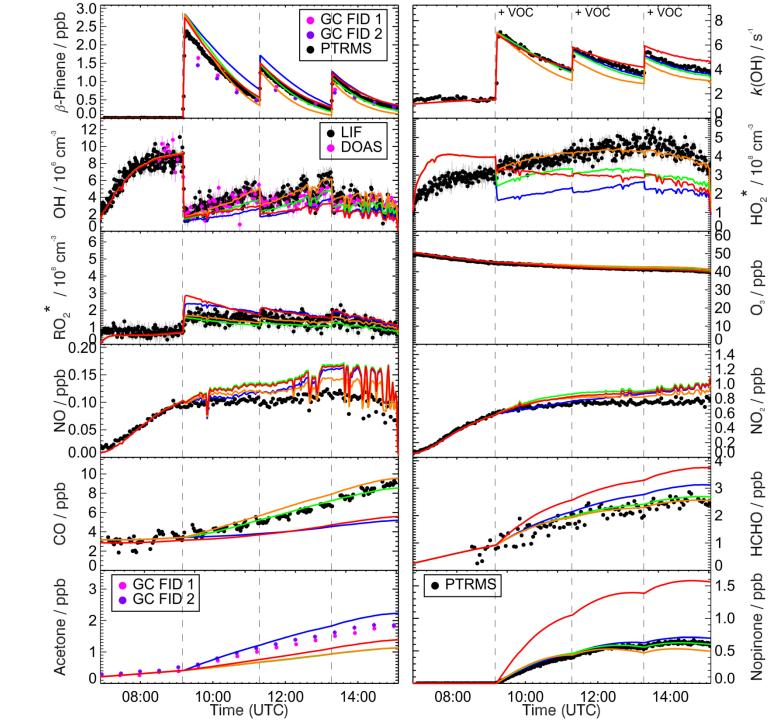
#### <u>MCM:</u>

- OH and HO<sub>2</sub> underpredicted
- Nopinone over-predicted

#### Vereecken and Peeters, 2012:

- Nopinone yield fixed
- Radical production still too low
- $\rightarrow$  OH and HO\_2 ratio consistent





## $\beta$ -pinene – OH budget

$$\begin{split} P(OH) &= j_{O(^{1}D)}[O_{3}] * 2f_{OH} \\ &+ j_{HONO}[HONO] + \\ &\propto k_{1}[VOC][O_{3}] + k_{2}[HO_{2}][NO] \\ &+ k_{3}[HO_{2}][O_{3}] \end{split}$$

$$D(OH) = k(OH) * [OH]$$

